TOTAL SYNTHESIS OF (+) - MILBEMYCIN β_1

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Abstract: The successful sulphone anion stabilised coupling of a monocyclic C-1 to C-10 unit (3) with the "northern hemisphere" C-11 to C-25 fragment (2) of the milbemycins produces a compound which may be further elaborated in fourteen steps to the macrocyclic natural product (+) - milbemycin β_I (1).

The milberrycins and avermectins¹ have attracted considerable chemical and biological interest over the past decade due to their potent antiparasitic and insecticidal properties. Synthesis in the area has been extensive, culminating in several excellent total syntheses of milberrycin $\beta_{3,2}$ the simplest member of the series, and the elegant construction of the structurally more complex avermectin A1a.³



Our work in this area was initiated by the need to define new methods for the preparation of the inherent spiroacetal unit;⁴ we also sought to devise alternative coupling sequences that would be applicable to the whole family of milbemycins and avermeetins. Previously we have reported an efficient synthesis of the "northern" C-11 to C-25 fragment (2) common to many milbemycins,^{5,6} the preparation of various monocyclic "southern" units such as 3, and several model studies for coupling and analogue preparation.⁶ Here we report the first synthesis of (+) - milbemycin $\beta_1(1)$,⁷ encompassing the methods that we have developed.



In an interesting coupling of the optically pure fragments 2 and 3^6 we found that the sulphone (3) may be reacted with 2 equivalents of t-butyl lithium to afford a dianion which specifically reacts through the C-10 carbon atom to give the hydroxy sulphone products (4) in excellent yield (83%).⁸ The conformation of the dianion derived from 3 is restricted such that the possibility of competing β -elimination is prohibited. We were unable to detect any other double bond isomers about the C-9 position during this coupling process. Reductive elimination of 4 using sodium amalgam in methanolic tetrahydrofuran in the presence of Na₂HPO₄ afforded the required E,E-diene (5) which was benzoylated in the usual way to give 6. Elaboration to the 16-membered macrolide required initial deprotection of the primary t-butyldiphenylsilyl ether using ⁿBu₄NF. Oxidation of the C-1 alcohol was accomplished using a two stage procedure; reaction with tetra-n-propylammonium perruthenate (TPAP)⁹ gave the aldehyde, which was then further oxidised to the carboxylic acid (7) using sodium chlorite.¹⁰ After removal of the benzoyl groups with sodium methoxide, the resulting dihydroxy acid (8) was cyclized to the macrolide $(9)^{11}$ using the excellent Mukaivama procedure.¹² Oxidation with TPAP and subsequent removal of the benzylidene acetal with moist trifluoroacetic acid afforded the ketone (10). For the remaining steps of the synthesis we chose to effect the introduction of the important 3,4-double bond by a syn-elimination sequence, since other workers have shown that difficulties arise in the related avermectins when unsaturation is present at an early stage in the synthesis, necessitating a less than satisfactory final deconjugation process, 3.13 Reaction of 10 with t-butyldimethylsilyl triflate to simultaneously protect the primary alcohol and form the thermodynamic enol ether was followed by treatment with phenylselenenyl chloride to give the protected ketoselenide (11). Oxidation of 11 to the corresponding selenoxides using the Davis oxaziridine reagent¹⁴ proceeded well, the subsequent elimination affording both the endo and exo products in a 2:1 ratio. The possibility of aromatisation of the endo- enone system made it most convenient to simply work-up the crude mixture with sodium borohydride and cerium (III) chloride¹⁵ to give the alcohols 12, 13 and 14 in a 1:1:1 ratio, in 81% combined yield. The alcohol (13) could then be recycled to 12 by oxidation (TPAP) and reduction as described above. Methylation of 12 with methyl iodide and silver (I) oxide under ultrasonication, followed by deprotection with HF/pyridine in acetonitrile, gave (+) milberrycin β_1 (1) in 55% overall yield. The synthetic sample¹⁶ of 1 was identical by ¹H nmr, IR, mass spectral, $[\alpha]_D$ and the comparisons (3 solvent systems) to authentic material kindly provided by the Sankvo Company.

We believe that the synthesis described above demonstrates the viability of our approach and coupling strategy for the preparation of this important class of antiparasitic agents.

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(iii) 2 eq. TBAF, THF, reflux, 91%; (iv) TPAP, NMO, 4Å ground sieves, CH₂Cl₂, 76%; (v) NaO₂Cl, 2-methyl-2-butane, KH₂PO₄, t-BuOH/H₂O; (vi) NaOMe, MeOH; (vii) 2-chloro-1-methyl pyridinium iodide, Et₃N, CH₃CN, reflux, 49% from aldehyde; (viii) TPAP,NMO, 4Å ground sieves, CH₂Cl₂, 83%; (ix) TFA, CH₂Cl₂, 92%; (x) 4 eq. TBDMSOTf, 20 eq. Et₃N, R.T., CH₂Cl₂, then PhSeCl, -78°C, 50%; (xi) 2-Benzenesulphonyl-3-(p-nitrophenyl)oxaziridine, CDCl₃, R.T., then NaBH₄, CeCl₃, MeOH, R.T.; (xii) TPAP, 4Å ground sieves, CH₂Cl₂, then NaBH₄, CeCl₃, MeOH, R.T.; (xiii) TPAP, 4Å ground sieves, CH₂Cl₂, then NaBH₄, CeCl₃, MeOH, R.T.; (xiii) MeI, Ag₂O,))), 73%; (xiv) HF, py, CH₃CN, 75%.

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References and Footnotes:

- 1. H.G. Davies and R.H. Green, Nat. Prod. Rep., 3, 87 (1986), and references therein.
- D.R. Williams, B.A. Barner, K. Nishitani, and J.G. Phillips, J. Am. Chem. Soc., 104, 4708 (1982); A.G.M. Barrett, R.A.E. Carr, S.V. Attwood, G. Richardson, and N.D.A. Walshe, J. Org. Chem., 51, 4840 (1986); P.J. Kocieński, S.D.A. Street, C. Yeates, and S.F. Campbell, J. Chem. Soc., Perkin Trans. I, 2171, 2183, 2189 (1987); R. Baker, M.J. O'Mahony, and C.J. Swain, J. Chem. Soc., Perkin Trans. I, 1623 (1987); S.R. Schow, J.D. Bloom, A.S. Thompson, K.N. Winzenberg, and A.B. Smith, III, J. Am. Chem. Soc., 108, 2662 (1986); M.T. Crimmins, D.M. Bankaitis-Davis, and W.G. Hollis, Jr., J. Org. Chem., 53, 652 (1988).
- S.J. Danishefsky, D.M. Armistead, F.E. Wincott, H.G. Selnick, and R. Hungate, J. Am. Chem. Soc., 109, 8117, 8119 (1987).
- 4. For a review of spiroacetal chemistry, see: A.F. Kluge, Heterocycles, 24, 1699 (1986).
- 5. C. Greck, P.Grice, S.V. Ley, and A. Wonnacott, Tetrahedron Lett., 27, 5277 (1986).
- N.J. Anthony, T. Clarke, A.B. Jones, and S.V. Ley, *Tetrahedron Lett.*, 28, 5755 (1987); C. Greck, P. Grice, A.B. Jones, and S.V. Ley, *Tetrahedron Lett.*, 28, 5759 (1987); N.J. Anthony, P. Grice, and S.V. Ley, *Tetrahedron Lett.*, 28, 5763 (1987).
- For details of isolation and structure determination, see: H. Mishima, M. Kurabayashi, C. Tamura, S. Sato, H. Kuwano, and A. Saito, *Tetrahedron Lett.*, 16, 711 (1975); Y. Takiguchi, H. Mishima, M. Okuda, M. Terao, A. Aoki, and R. Fukuda, J. Antibiot., 33, 1120 (1980).
- 8. All new compounds gave satisfactory spectral, microanalytical and/or accurate mass data.
- 9. W.P. Griffith, S.V. Ley, G.P. Whitcombe, and A.D. White, J. Chem. Soc., Chem. Commun., 1625 (1987).
- 10. G.A. Kraus and M.J. Taschner, J. Org. Chem., 45, 1175 (1980).
- Data for compound (9): [α]²⁰_D = + 3.8° (c = 1.1, CHCl₃); δ_H (500MHz, CDCl₃) 7.63 (2H, m, Ph), 7.39-7.32 (3H, m, Ph), 6.03 (1H, dd, J 15.2, 10.3 Hz, H-10), 5.97 (1H, s, PhCH), 5.90 (1H, d, J 10.2 Hz, H-9), 5.74 (1H, dd, J 15.3, 5.1 Hz, H-11), 5.46 (1H, tt, J 11.4, 4.6 Hz, H-19), 4.87 (1H, br.d, J 11.2 Hz, H-15), 4.84 (1H, d, J 13.8 Hz, H-8'), 4.54 (1H, d, J 13.8Hz, H-8'), 3.67 (1H, m, H-17), 3.50 (1H, dt, J 11.5, 4.7 Hz, H-5), 3.25 (1H, dq, J 9.8, 6.2 Hz, H-25), 3.04 (1H, dd, J 13.8, 4.2 Hz, H-6_{eq}), 2.90 (1H, dd, J 12.4, 3.6 Hz, H-2), 2.57 (1H, m, H-12), 2.38-2.17 (3H, m), 2.05-1.68 (6H, m) 1.67 (3H, s, Me-14), 1.57-1.12 (6H, m), 1.09-1.05 (6H, m), 1.03 (3H, d, J 6.4 Hz, Me), 1.00-0.84 (3H, m), and 0.82 (3H, d, J 6.6 Hz, Me); m/z (FAB from 3-nitrobenzyl alcohol) 621 (MH⁺), 602 (M⁺-H₂O), and 515 (MH⁺-PhCHO); (Observed MH⁺, 621.3791. Calc. for C₃₈H₅₃O₇ MH, 621.3791).
- 12. T. Mukaiyama, M. Usui, and K. Saigo, Chem. Lett., 49 (1976).
- S. Hanessian, A. Ugolini, D. Dubé, P.J. Hodges, and C. André, J. Am. Chem. Soc., 108, 2776 (1986);
 B. Fraser-Reid, H. Wolleb, R. Faghih, and J. Barchi, Jr., J. Am. Chem. Soc., 109, 933 (1987).
- 14. F.A. Davis, O.D. Stringer, and J.M.Billmers, Tetrahedron Lett., 24, 1213 (1983).
- 15. J.-L. Luche, J. Am. Chem. Soc., 100, 2226 (1978).
- 16. Data for compound (1): [α]²⁰_D=+129.1^o (c = 0.3, CHCl₃); δ_H (500MHz, CDCl₃) 6.41 (1H, d, J 11.3 Hz, H-9), 6.25 (1H, dd, J 14.7, 11.3 Hz, H-10), 5.45 (1H, dd, J 14.7, 9.8 Hz, H-11), 5.39 (1H, tt, J 11.5, 4.7 Hz, H-19), 5.25 (1H, m, H-3), 4.87 (1H, m, H-15), 4.26 (1H, dd, J 12.1, 6.1 Hz, H-8), 4.17 (1H, dd, J 12.1, 4.6 Hz, H-8), 4.04 (1H, br m, H-5), 3.78 (1H, d, J 1.6 Hz, OH-7), 3.55 (1H, m, H-17), 3.48 (1H, m, H-2), 3.36 (3H, s, MeO-5), 3.24 (1H, dq, J 9.7, 6.2 Hz, H-25), 2.47 (1H, m, H-12), 2.27-2.16 (3H, m, H₂-16 and H-13), 1.93-1.75 (5H, m), 1.80 (3H, s, Me-4), 1.64 (1H, m), 1.59 (3H, s, Me-14), 1.56-1.37 (5H, m), 1.25-1.19 (1H, m, H-20), 1.11 (3H, d, J 6.3 Hz, Me-25), 1.03 (3H, d, J 6.3 Hz, Me-12), 0.82 (3H, d, J 6.6 Hz, Me-24), and 0.77 (1H, q, J 11.6 Hz, H-18_{ax}).

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